REMARKS

Claims 1-15 are all the claims pending in the application. Claims 1 and 2 have been

amended to make an editorial change.

Entry of the above amendment is respectfully requested.

Anticipation Rejection over Mikito et al.

On page 2 of the Office Action, in paragraph 3, claims 1-9 and 10-15 are rejected under

35 U.S.C. 102(b) as being anticipated by Mikito et al. (JP 10-208730, abstract and Computer

translation).

In response, Applicant notes initially that metallic lithium means lithium of which the

oxidation number is zero. Therefore, metallic lithium does not include a lithium-containing host

material like graphite or a lithium-containing composite oxide like Li<sub>1+x</sub>Mn<sub>2</sub>O<sub>4</sub>.

Applicant's claims 1 and 2 recite that metallic lithium is used as a negative active

material. However, the negative active material described in Mikito et al is a host material which

is capable of storing and releasing <u>lithium ions</u>, like a graphite or carbon, **not** <u>metallic</u> lithium.

Applicant wishes to emphasize that lithium ion in the host material is absolutely not the metallic

lithium of the present claims 1 and 2. Therefore, Applicant submits that it is clear that present

claims 1 and 2 are novel over Mikito et al.

The Examiner describes in page 2, item 4 of the Office Action dated October 6, 2003,

"Mikito et al. teaches a non-aqueous electrolyte secondary battery comprising a positive

electrode comprising a lithium metal oxide such as  $Li_{1+x}Mn_2O_4$  ( $0 \le x \le 1$ ) and a material having

5

Attorney Docket No.: Q65355

lithium ion release potential at a lower potential than a lithium ion release potential of the lithium

metal oxide". However, this description does not show that Mikito et al. teaches metallic lithium

electrically connected to the positive electrode, because metallic lithium means lithium of which

the oxidation number is zero. Therefore, Applicant submits that it is clear that present claims 1

and 10 are novel over Mikio et al.

Applicant's claims 10 and 11 refer to metallic lithium on a negative electrode. Again,

Applicant submits that it is therefore clear that present claims 10 and 11 are novel over Mikito et

al.

In view of the above, Applicant submits that the present claims are not anticipated by

Mikito et al. Accordingly, withdrawal of this rejection is respectfully requested.

Anticipation Rejection over Yukio et al.

On page 3 of the Office Action, in paragraph 5, claims 1, 3, 9-10, and 14 are rejected

under 35 U.S.C. 102(b) as being anticipated by Yukio et al. (JP 5-144471, abstract).

In response, Applicant notes that the Examiner describes in page 3, lines 3-5 of the Office

Action dated October 6, 2003, that "Yukio et al. teaches a secondary battery with nonaqueous

electrolyte comprising ... a negative electrode made of carbon material" (emphasis added).

However, Applicant's claim 1 recites that metallic lithium is used as a negative active material,

and Applicant's claim 10 refers to metallic lithium on a negative electrode. Carbon material is

absolutely not the metallic lithium of present claims 1 and 10. Therefore, Applicant submits that

it is clear that present claims 1 and 10 and their dependent claims are novel over Yukio et al.

6

AMENDMENT UNDER 37 C.F.R. § 1.111

U.S. Application No.: 09/899,208

Attorney Docket No.: Q65355

In view of the above, Applicant submits that the present claims are not anticipated by

Yukio et al. Accordingly, withdrawal of this rejection is respectfully requested.

Conclusion

In view of the above, reconsideration and allowance of this application are now believed

to be in order, and such actions are hereby solicited. If any points remain in issue which the

Examiner feels may be best resolved through a personal or telephone interview, the Examiner is

kindly requested to contact the undersigned at the telephone number listed below.

The USPTO is directed and authorized to charge all required fees, except for the Issue

Fee and the Publication Fee, to Deposit Account No. 19-4880. Please also credit any

overpayments to said Deposit Account.

Respectfully submitted,

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Date: January 6, 2004

7